Objective Paper Code

## Intermediate Part Second - 103

## CHEMISTRY (Objective)

Time: 20 Minutes

GROUP - I Marks: 17

Roll No.:

8487 You have four choices for each objective type question as A, B, C and D. The choice which you think is correct, fill the relevant circle in front of that question number on computerized answer sheet. Use marker or pen to fill the circles. Cutting or filling two or more circles will result in zero marks in that question. Attempt as many questions as given in objective type question paper and leave other circles blank.coa

S.#	Questions	A	В	С	D
1	During nitration of benzene, the active nitrating agent is:	NO <sub>2</sub>	HNO <sub>3</sub>	NO	NO <sub>2</sub> <sup>+</sup>
2	For which Mechanism, the first step involved is the same?	E1 and E2	E2 and S <sub>N</sub> 2	S <sub>N</sub> 1 and E2	E1 and S <sub>N</sub> 1
3	Which enzyme is not involved in fermentation of starch?	Diastase	Zymase	Urease	Invertase
4	Cannizzaro's reaction is not given by:	Acetaldehyde	Formaldehyde	Benzaldehyde	Trimethyl acetaldehyde
5	Acetic acid was first isolated from:	Milk	Vinegar	Butter	Red ant
6	Which is an addition polymer?	Nylon 6-6	Terylene	Polystyrene	Epoxy resin
7	Phosphorus helps the growth of:	Root	Leave	Stem	Seed
8	The pH range of the acid rain is:	7-6.5	6.5 – 6	6-5.6	Less than 5
9	Half of the atmospheric mass is concentrated in the lower:	5.6 km	10.6 km	15.6 km	20.6 km
10	Among alkali metal ions, minimum hydration energy is shown by:	Li⁺	Na <sup>+</sup>	Rb <sup>+</sup>	K <sup>+</sup>
11	The oxide of beryllium is:	Acidic	Basic	Neutral	Amphoteric
12	Which element is not present abundantly in earth's crust?	Si	Αℓ	Na	0
13	Which catalyst is used to contact process?	V <sub>2</sub> O <sub>5</sub>	Fe <sub>2</sub> O <sub>3</sub>	SO <sub>3</sub>	Ag <sub>2</sub> O
14	Chlorine heptaoxide ( $C\ell_2O_7$ ) reacts with water to form:	Hypochlorous acid	Chloric acid	Perchloric acid	Chlorine and oxygen
15	Coordination number of Pt in [Pt Cl(NO <sub>2</sub> )(NH <sub>3</sub> ) <sub>4</sub> ] is:	02	04	01	06
16	Which set of hybrid orbitals has planar triangular shape?	sp <sup>2</sup>	sp <sup>3</sup>	sp	dsp <sup>2</sup>
17	Vinyl acetylene combines with HCℓ to form:	Poly acetylene	Chloroprene	Benzene	Divinyl acetylene