/ No	( To be filled in by the	
(Academic Sessions 2019 – 2021 to 2021 – 2023)		
HEMISTRY 223-1 <sup>st</sup> Annual-(INTER PART – II) Time Allowed: 20 Minutes		
Į.PAPI		ximum Marks: 17
PAPER CODE = 8486 (6.14 ON A DATE OF THE PAPER OF THE PAP		
Note: Four possible answers A, B, C and D to each question are given. The choice which you think is correct,		
fill that circle in front of that question with Marker or Pen ink in the answer-book. Cutting or filling two or more circles will result in zero mark in that question.		
1-1 Elimination bimolecular reactions involve:		
1	1	
	(A) First order kinetics (B) Second order kinetics	
	(C) Third order kinetics (D) Zero order kinetics	
2	Which one of the following is a typical transition element:	
	(A) Sc (B) Y (C) Zn	(D) Co
3	Cannizzaro's reaction is given by:	
	(A) Formaldehyde (B) Acetone (C) Acetaldehyde	(D) 3-pentanone
4	Which one of the following does not belong to alkaline earth metals	:
	(A) Be (B) Ra (C) Ba	(D) Rn
5	The normal amount of overhead ozone is about:	
	(A) 150 DU (B) 250 DU (C) 350 DU	(D) 450 DU
6	Synthetic rubber is made by the polymerization of:	(B) 430 BO
	(A) Chloroform (B) Acetylene (C) Divinyl acetylene	(D) Chloroprene
7	Which catalyst is used in the contact process:	(b) Chloroprene
	(A) $Fe_2O_3$ (B) $V_2O_5$ (C) $SO_3$	(D) 40 O
8	Which statement is incorrect:	(D) Ag <sub>2</sub> O
1	(A) All the metals are good conductor of electricity (B) All the metals a	are good conductor of heat
-	(C) All the metals form positive ions (D) All the metals	form acid oxides
9	Which compound is called a universal solvent:	
	(A) $H_2O$ (B) $CH_3OH$ (C) $C_2H_5OH$	(D) $CH_3 - O - CH_3$
10	Which one is argillaceous material:	, , , , , , ,
	(A) Clay (B) Lime (C) Marble	(D) Marine shell
11	The electrophile in aromatic sulphonation is:	(D) Warme shell
		(D) got
12		(D) $SO_3^+$
12	The anhydride of $HC\ell O_4$ is:	
12	(A) $C\ell O_3$ (B) $C\ell_2 O_5$ (C) $C\ell O_2$	(D) $C\ell_2O_7$
13	Which of the following element is present in all amino acids:	
	(A) Br (B) N (C) $Cl$	(D) Cu
14	Ethers show the phenomenon of:	
	(A) Position isomerism (B) Functional group isomerism	
	(C) Metamerism (D) Cis-trans isomerism	
15	Aluminium oxide is:	,
	(A) Acidic oxide (B) Basic oxide (C) Amphoteric oxide	(D) None of these
16	According to Lewis concept ethers behave as:	(2) Hone of these
	(A) Acid (B) Base (C) Acid as well as a base	(D) None of these
17	Alkane nitriles can be converted into carboxylic acids by :	(2) Troile of these
	(A) Hydration (B) Acid hydrolysis (C) Hydrogenation	(D) Oxidation