Roli 1					
	(Acad MISTRY CR – II (Essay Type)	demic Sessions 2018 – 2020 to 2020 - 222-(INTER PART – II) GROUP – II	Time Allowed: 2.40 hours Maximum Marks: 68		
	•	SECTION - I			
		any EIGHT (8) questions:	16		
(i	<ul> <li>Lanthanide contract Give reason.</li> </ul>	tion controls the atomic sizes of eleme	ents of 6 <sup>th</sup> and 7 <sup>th</sup> periods.		
(ii	) Explain the variatio	ons in melting points along the short p	eriods.		
(iii	) Why is the aqueous	s solution of $Na_2CO_3$ alkaline in natural	re? Give reaction only.		
(iv	) What happens wher	n LiOH is heated to red hot? 'Give rea	ection.		
(v	) How does boric acid	d react with: (a) Ethyl alcohol	(b) $Na_2CO_3$		
(vi	) Give chemistry of b	porax-bead test.			
(vii	) Name the allotropes	s of phosphorus. How red phosphorus	s is prepared?		
(viii	) Give the reaction of	f phosphorus with: (a) Thionyl chlo	oride (b) $C\ell_2(g)$		
(ix	) Briefly discuss the j	property of paramagnetism in transition	on elements compounds.		
		erstitial compounds. (b) Substitution			
(xi	) Briefly discuss the	digestion process in paper manufactur	ring.		
(xi	) Discuss the reaction	ns taking place in first 24 hours of sett	ting of cement.		
3. W	rite short answers to	any EIGHT (8) questions:	16		
(i	) Why HF is weaker	acid than HCL?			
(ii	) What are freons and	d teflon?			
(ii	i) Discuss carbonizati	ion of coal.			
(iv	) Define functional g	roup. Give one example.			
(v	) Why is sigma bond	inert?			
(vi	) Discuss hydroxylati	ion of ethene.			
(vi	) How is water added	d to propyne? Write reaction.			
(vii	i) How are tetramethy	yl and tetraethyl lead prepared?	,		
(ix	) How can 1-chlorop	ropane be converted to propene?			
(x	) Define polysacchar	ides, also give example.			
(x	i) Differentiate between	en DNA and RNA.			
(xi	i) What is hardening	of oils? Give reaction.			
4. V	rite short answers to	any SIX (6) questions:	12		
(	i) What is Wurtz-Fitti				
(i		c aromatic hydrocarbons? Give two			
(ii		. (., -, ,	Cyclohexanol.		
(i		r the preparation of ethanol by the ferr	mentation of Molasses.		
(	v) What is iodoform to	est?	(Turn Over)		
			(Iuni Over)		

- 4. (vi) Convert acetylene into acetic acid.
  - (vii) Write down the structural formulas of Glycine and Alanine amino acids.
  - (viii) How detergents are threat to aquatic animal life?
  - (ix) Write down the four harmful effects of acid rain.

## SECTION - II

## Note: Attempt any THREE questions.

5.	(a)	Write down two points of similarities and two points of differences between hydrogen and halogen (VII A)	
	(b)	Give one method for the preparation of $H_3BO_3$ . How does it react with: (i) NaOH (ii) $Na_2CO_3$ (iii) $C_2H_5OH$	4
6.	(a)	Discuss hard finish plasters and cement plaster.	4
	(b)	How steel is manufactured by Bessemer process.	4
7.	(a)	Discuss structure of ethyne on the basis of sp-hybridization.	
	(b)	Convert ethyl bromide into: (i) Ethane. (ii) Propane. (iii) Ethyl alcohol (iv) n-Butane	4
8.	(a)	Give the reaction of ethene with (i) $H_2SO_4$ (ii) $O_3$ (iii) HOX (iv) $Br_2$	4
	(b)	Describe the various tests for identification of carboxyl compounds (Any four ).	4
9.	(a)	Write the mechanism of nitration and sulphonation of benzene.	4
	(b)	Describe the two reactions of alcohol and phenol in which "O-H" bond break.	4

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